# Universida<sub>de</sub>Vigo

## Subject Guide 2022 / 2023

*		Jubje	ct Guide	2022   2023
IDENTIFYIN				
	cience and technology			
Subject	Materials science			
	and technology			
Code	V12G363V01301			
Study	Grado en			
programme	Ingeniería en Tecnologías			
<del></del>	Industriales			
Descriptors	ECTS Credits Choose Year			nester
Teeshing	6 Mandatory 2nd		1st	
Teaching	#EnglishFriendly			
language	Spanish Galician			
Department	Galician			
Coordinator	Figueroa Martínez, Raúl			
Coordinator	Pena Uris, Gloria María			
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General	The aim of this subject is to introduce the main concepts of materials technology as	s well as	s to stud	у
description	applications of the most common materials			
adapt to B4 CG4 Ab transmi B6 CG6 Ca C9 CE9 Kno betwee D1 CT1 Ana D5 CT5 Info D9 CT9 Ap	owledge of basic and technological subjects that enable students to learn new metho o new situations. ility to solve problems through initiative, decision-making, creativity, critical reasonin t knowledge, skills and abilities in the field of industrial engineering. pacity for handling specifications, regulations and mandatory standards. owledge of the fundamentals of the science, technology and chemistry of materials. I n microstructure, the synthesis, processing and properties of materials. alysis and synthesis. ormation Management. olication of knowledge. elf learning and work.	ng, and	to comm	nunicate and
Learning ou				
	ults from this subject	Tra	Resi	d Learning ults
types of mat		B3	C9	D10
thermal and	he relationship between microstructure and properties (mechanical, electrical, magnetic) in a material	B3	C9	
	he mechanical performance of metallic, ceramic, plastic and composite materials.	B4 B6		
Know the post thermal treat	ssibilities of modification of material properties through mechanical processing and tment	B4	C9	D9
Know the ma	in techniques for materials characterization	B3 B6	C9	
Acquire abili	ties in handling materials diagrams and charts	-		D1
	ties in undertaking standardized tests on materials, under supervision	B6	C9	D10

Contents			
Торіс			
Introduction	Introduction to the Science and Technology of Material. Classification of the materials. Terminology. Orientations for the follow-up of the matter.		
Crystalline arrangement.	Crystalline and amorphous solids. Crystalline lattices, characteristics and imperfections. Allotropic transformations.		
Properties of materials. Laboratory practices.	Mechanical, chemical, thermal, electric and magnetic properties. Standars for materials analysis. Compressive and tensile deformation. Principles of fracture mechanisms. Toughness. Hardness. Main test methods. Introduction to metallography. Binary isomorphous and eutectic systems. Microstructure in eutectic alloys. Analyses of practical situations.		
Metallic materials.	Solidification. Constitution of alloys. Grain size. Main binary phase diagrams. Processing. Carbon steels: classification and applications. Cast iron alloys. Heat treatments: aims, fundamentals and classification. Annealing, normalizing, quenching and tempering. Nonferreous alloys.		
Plastic materials	Classification accoording to the molecular structure: Thermoplastics, thermosets and elastomers. Properties and assessing methods. Forming processes. Introduction to the Composite Materials.		
Ceramic materials	Classification and properties. Glasses and traditional ceramics. Technical Ceramics. Cements: phases, types and main applications. Concrete. Processing of ceramic materials.		

Planning			
	Class hours	Hours outside the classroom	Total hours
Introductory activities	1	0	1
Lecturing	31	56	87
Laboratory practical	16.75	18	34.75
Autonomous problem solving	0	12.2	12.2
Mentored work	0.5	9	9.5
Problem and/or exercise solving	1.5	0	1.5
Presentation	0.25	0	0.25
Report of practices, practicum and external practices 0		2	2
Self-assessment	0	0.3	0.3
Objective questions exam	1.5	0	1.5
*The information in the planning table is for	or guidance only and does no	t take into account the het	erogeneity of the students.

Methodologies	
	Description
Introductory activities	A presentation of the course is made: contents, organization, methodologies to be used, schedule and evaluation system. Emphasis is placed on student participation and the personalized tutoring system.
Lecturing	During the academic course, the teacher exposes the main contents of the course, encouraging the active participation of the students. Exercises and type problems are solved, and hands on science methodology will be also applied.
Laboratory practical	Activities for the practical application of the knowledge acquired in the theoretical sessions. They are performed in the laboratory with specialized equipment and in accordance with applicable standards
Autonomous problem solving	Throughout the course, students will be offered different set of problems and questions that they will have to solve by themselves, demonstrating the capacity for learning and developing autonomous work.
Mentored work	The instructor will propose several projects to be carried out in small groups. The projects with be related to the characterization of materials commonly used in technological applications. Students must complete a revision of the literature concerning to the topic of the project, revise the existing standards and other sources of information. Finally, the project must be exposed to the instructor and to their classmates.

Methodologies

Description

The teacher will guide and resolve any doubts that the student may have in relation to the contents explained in the lectures.
The laboratory teacher will guide the students in the development of the practical classes, clarifying their doubts and guiding them to achieve the best understanding of the practical classes
During the development of the tasks proposed to be done in small groups, the students will have the guidance and help of the teacher
Description
The students will have the support of the teacher to solve the doubts that can arise in the resolution of the numerical problems proposed in class, as well as those that are offered for their autonomous work.
The laboratory teacher will guide the students in the resolution of the questions formulated in the practical classes and will help in the doubts that may arise in the writing of the practical reports.
The teacher will design the self-assessment tests that the student must take throughout the course, and will guide the students in their completion, solving the technical questions that may arise

Assessment		0 11/2 11	_		
	Description	Qualification	L		ning
Laboratory practical	The attendance and active participation of the student in the practical sessions will be valued	1	B3 B6	C9	D1 D9 D10
Problem and/or exercise solving	Student learning in practical sesions will be evaluated by means of a written exam, which will include of exrcices and problems (7%) The final exam will include of problems and exercises similar to those raised during the course (35%)	42	B4 B6	C9	D1 D9 D10
Presentation	The projects will be assessed after the oral exposition. These are the items to be taken into account for the assessment: revised literature, structure of the contents used in the presentation and ability to reply to the comments given by the instructor and/or classmates.		B4 B6	C9	D1 D5 D10
Report of practices, practicum and external practices	The student must present a report of the practical sessions which will include the results obtained in the mechanical tests as well as the answers to the questions asked.	4	B6	C9	D9
Self-assessment	Resolution of proposed online questionnaires, which will consist of true and false questions and multiple choice questions	4	B3	C9	D9 D10
Objective questions exam	Student learning in practical sesions will be evaluated by means of a written exam, which will include of short answer questions and test questions (7%) The final exam will include hort answer questions and test questions (35%)	42	B3 B4	C9	D1 D5 D9 D10

### Other comments on the Evaluation

**Continuous assessment:** The continuous assessment activities will be carried out during the teaching period and correspond to 30% of the grade.

Final Exam: counts for 70% of the course grade. The exam will be taken on the official date set by the EEI direction.

#### **Requirements to pass the course:**

It is necessary to achieve a minimum score of 40% in the final exam, that is 2.8 / 7.

If this minimum is not reached, the course will be considered as not passed and, although the sum of the exam grade and the continuous evaluation is higher than 5, the maximum grade that will be included in the academic records will be 4.5 points.

**Renouncing continuous assessment:** Students that do not follow the continuous assessment activities, after receiving authorization from the EEI direction, will be evaluated through a final exam on the contents of all the course, covering both lecture and labo contents, counting for 100% of the grade. A minimum mark of 5 (50%) will be required to pass the course.

July exam (2nd Edition): In the July edition, the continuous assessment marks will be also considered (only marks

obtained in the current academic year). The characteristics of the exam will be the same as the first edition, and will be taken on the official date set by the EEI direction. Further in the July edition, the student can choose to be evaluated through a final exam on the contents of all the course, covering both lecture and labo contents, counting for 100% of the grade. A minimum mark of 5 (50%) will be required to pass the course. The student must notify the teacher of their choice well in advance.

**Extraordinary Call:** The extraordinary call exam contents will cover the entire course, including both lecture and labo contents, counting for 100% o the grade. A minimum mark of 5 (50%) will be required to pass the course.

**Ethical commitment:** Students are expected to carry out their work in accordance with an appropriate ethical behaviour. If the professor detects a behaviour that constitutes academic dishonesty (cheating, plagiarism, use of unauthorized electronic devices, for example) the student will be deemed not to meet all the criteria to pass the course, and will be informed that the final grade of this course will be FAIL (0.0). The use of any electronic device will not be allowed during the evaluation tests, unless expressly authorized. Introducing an unauthorized electronic device into the exam room will be considered reason enough for not passing the course in the present academic year, and the final grade will be: FAIL (0.0).

# Attention: If there is any mismatch between the contents of the 3 language versions of this teaching guide, those included in the Spanish version will be considered valid.

### Sources of information

Basic Bibliography Callister, William, Ciencia e ingeniería de los materiales, 2ª, Reverté, 2016

Askeland, Donald R, Ciencia e ingeniería de materiales, 6ª, Cengage Learning, 2012

Shackelford, James F, Introducción a la ciencia de materiales para ingenieros, 7ª, Pearson Educación, 2010 Complementary Bibliography

Smith, William F, Fundamentos de la ciencia e ingeniería de materiales, 5ª, McGraw-Hill, 2010

AENOR, Standard tests,

Montes J.M., Cuevas F.G., Cintas J., **Ciencia e ingeniería de los materiales / J.M. Montes, F.G. Cuevas, J. Cintas**, 1ª, Paraninfo, 2014

### Recommendations

Subjects that continue the syllabus Materials engineering/V12G380V01504

### Subjects that are recommended to be taken simultaneously

Fundamentals of manufacturing systems and technologies/V12G380V01305 Fluid mechanics/V12G380V01405 Thermodynamics and heat transfer/V12G380V01302

### Subjects that it is recommended to have taken before

Computer science: Computing for engineering/V12G350V01203 Physics: Physics I/V12G380V01102 Physics: Physics II/V12G380V01202 Mathematics: Algebra and statistics/V12G380V01103 Mathematics: Calculus I/V12G380V01104 Chemistry: Chemistry/V12G380V01205