## Universida<sub>de</sub>Vigo

Subject Guide 2018 / 2019

IDENTIFYIN	<u> </u>			
	tical chemistry			
Subject	Pharmaceutical			
	chemistry			
Code	V11G200V01903	,		
Study	(*)Grao en Química			
programme				
Descriptors	ECTS Credits	Choose	Year	Quadmester
	6	Optional	4th	2nd
Teaching	Spanish			
language				
Department	Chemical Engineering			
	Organic Chemistry			
Coordinator	Terán Moldes, María del Carmen			
Lecturers	Álvarez Álvarez, María Salomé			
	Terán Moldes, María del Carmen			
E-mail	mcteran@uvigo.es			
Web				
General	The matter is allocated to contribute to the st	udents basic knowledge	s on Pharmaceu	tical Chemistry, an
description	interdisciplinar science that is among differen			
•	study of the bioactive compounds and in parti			
	of action at molecular level.	<b>,</b> ,	•	

#### Competencies

Code

- Al Students have demonstrated knowledge and understanding in a field of study that builds upon their general secondary education, and is typically at a level that, whilst supported by advanced textbooks, includes some aspects that will be informed by knowledge of the forefront of their field of study
- A3 Students have the ability to gather and interpret relevant data (usually within their field of study) to inform judgments that include reflection on relevant social, scientific or ethical issues
- A4 Students can communicate information, ideas, problems and solutions to both specialist and non-specialist audiences
- A5 Students have developed those learning skills that are necessary for them to continue to undertake further study with a high degree of autonomy
- C19 Apply knowledge and understanding to solve basic problems of quantitative and qualitative nature
- C20 Evaluate, interpret and synthesize data and chemical information
- C22 Process and perform computational calculations with chemical information and chemical data
- C23 Present oral and written scientific material and scientific arguments to a specialized audience
- D1 Communicate orally and in writing in at least one of the official languages of the University
- D3 Learn independently
- D4 Search and manage information from different sources
- D5 Use information and communication technologies and manage basic computer tools
- D7 Apply theoretical knowledge in practice
- D8 Teamwork
- D9 Work independently
- D10 Work at a national and international context
- D12 Plan and manage time properly
- D13 Make decisions
- D14 Analyze and synthesize information and draw conclusions
- D15 Evaluate critically and constructively the environment and oneself
- D16 Develop an ethical commitment
- D17 Develop concern for environmental aspects and quality management

#### Learning outcomes

Expected results from this subject

Training and Learning Results

Diferenciate and understand the concepts: drug, active principle, medicine and pharmacological target	A4	C20 C23	D1 D4 D5 D14
Differentiate the types of receptors, as well as an agonist drug from an antagonist.	A4 A5	C20 C23	D1 D3 D4 D5 D7 D9 D13 D14
Relate the physical chemical properties of drugs with their pharmacokinetics.	A1 A3 A5	C19 C20 C22 C23	D1 D3 D5 D7 D8 D14
Differentiate the pharmacomodulation techniques.	A3 A5	C19 C20 C23	D1 D4 D5 D7 D8
Differentiate a chemoterapeutic from a pharmacodynamic agent	A3 A4 A5	C19 C20 C23	D1 D3 D4 D7 D9
Familiarise with the most recent tools in drug design: combinatorial chemistry and computer-aided drug design (QSAR and docking methods)	A3 A5	C19 C20 C22 C23	D1 D3 D4 D5 D8 D12 D13 D15 D16
Describe the methods of structural analysis involved in drug design and differentiate the type of information that they provide	A3 A5	C19 C20 C22 C23	D1 D3 D5 D7 D9 D14 D15
Identify the different forms of drug administration and their fundamentals.	A1 A3 A4 A5	C19 C20 C23	D1 D3 D4 D9 D14
Identify the formulation and composition variables in the preparation of suspensions and emulsions, and describe their characteristic properties, as well as and the instability phenomena	A3 A5	C19 C20 C23	D1 D3 D9 D13 D14
Recognise the main stages of fermentative and enzymatic processes applied to the drug production, including production and purification steps	A3 A5	C19 C20 C22 C23	D1 D3 D4 D7 D8 D12 D14 D15

Apply the basic principles of safety and pollution control in operations and processes oriented to drug production	A3 A5	C19 C20 C23	D1 D3 D5 D8 D10 D13	
			D16 D17	
Explain the sampling, pretreatment and sample preparation, as well as the	A3	C19	D1	_
appropriate instrumental techniques for the analysis of prime matters, bioactive compounds and	A5	C20	D3	
pharmaceutical formulations in the biological media		C22	D8	
		C23	D13	
			D14	

Contents	
Topic	
Subject 1. Introduction: general aspects of Pharmaceutical Chemistry	Definitions, aims and scope of the Pharmaceutical Chemistry. Nomeclature of drugs and classification systems. Chemotherapeutic and pharmacodynamic agents
Subject 2. Drug targets	Types of drug targets. Drug-target interactions. Nucleic Acids, enzymes and proteins as drug targets.
Subject 3. Receptors as drug targets	Types of receptors. Agonist, antagonist and inverse agonist drugs.  Measure and expression of pharrmacological effect. Drug tolerance and tachyphylaxis
Subject 4. Pharmacokinetic and related aspects	Absorption and transport through biological membranes, the Lipinski rules, bioavailabilty. Metabolism, prodrugs. Excretion. ways of drug administration and pharmaceutical forms.
Subject 5. Discovery, design and development of drugs	Strategies for lead discovery, serendipity, systematic screening, rational design. Pharmacomodulation. Patents. preclinical and clinical trials. Chemical development.
Subject 6. Strategies for drug design	Molecular modeling, indirect methods (QSAR, pharmacophore design), direct methods (docking).
Subject 7. Preparation, analysis and purification of drugs	Production in the pharmaceutical industry. fermentative processes. Drug processing.

Planning			
	Class hours	Hours outside the classroom	Total hours
Lecturing	26	52	78
Seminars	13	39	52
Studies excursion	3	3	6
Short answer tests	1	3	4
Essay questions exam	2	8	10

<sup>\*</sup>The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

Methodologies	
	Description
Lecturing	In these sessions the professor will present in a structured form the general contents of the program, doing emphasis in important or difficult aspects of the subject. In addition, the professor, in advance and through the Tem@ platform, will make available to the student the material that will be used in these sessions. Students should previously check and complete this material by using the recommended literature.  In addition, periodic controls will be carried out to follow the study and understanding of the subject. These tests will be performed during some master sessions which will be determined in advance
Seminars	They will devote time to discuss the most complicated aspects of the treated subjects, to use programs of molecular modeling which will allow to work with several biomolecules cocrystallized with different ligands, as well as to present works, researchs or summaries carried out by the students and related with the content of the subject.
Studies excursion	The students will visit a company of the pharmaceutical sector, in which it will be able to appreciate the process of production in all its phases.  After the visit the students will have to answer, in schedule of class, to a test related with this visit.

Per	rsoi	nali:	zed	attent	tion

**Methodologies Description** 

Seminars

Time devoted by the teachers to attend the needs and queries of the students related with the study of the subject and developed activities. The teachers will inform in the presentation of the subject about the available schedule.

Assessment					
	Description	Qualification		Training arning	g and Results
Lecturing	Contents developed in the program study will be evaluated by means of verbal or written questions formulated in the theoretical sessions. The writter questions will be referents to the content treated in the previous two or three weeks.		A1 A3	C19 C23	D14 D15 D16
Seminars	Attendance and participation in the sessions, exercices and questions resolution, as well as the presentation of reports, summaries and works, will be qualified	20	A1 A3 A4 A5	C19 C20 C22 C23	D1 D3 D4 D5 D7 D8 D9 D10 D12 D13 D14 D16
Studies excursion	Attendance and active participation in the visit, as well as the results of the test will be qualified.	10	_ A3	C20	D14 D15 D17
Short answer tests	A short exam (one hour long) will be carried out at week nine. In this exam will enter the subject explained until that moment.	15	A1 A3 A5	C19 C20	D7 D12 D13 D14
Essay question exam	ns A global exam will be carried on closing date of evaluation in order to analyze the adquired competencies	50	A1 A3 A5	C19 C20	D7 D12 D13 D14

#### Other comments on the Evaluation

Participation of students in any of the evaluation parts, such as attendance to seminars (four or more) or the performace of written exams, will involve the condition of presented and therefore the obtaining of a qualification. Students should have a minimum mark in some of the evaluation parts in order to pass the subject (5 or more points). This minimum mark should be of 4 in the global exam, as well as in seminars and study visit.

Evaluation in the July Call

1. Mark obtained by the students during the academic course: maximum 3.5 points

Marks obtained from verbal or written questions formulated in the theoretical sessions (maximun 0.5 point), visit test (maximun 1 point) and seminars (maximum 2 points) will be preserved.

2. Work carried out by the students: maximum 1.5 points

Finished the evaluation process of June, teachers will propose to the students who have not pass the subject to perform an individual work in order to adquire the competencies of which they will be evaluated in July. This work should be presented and defended before the exam.

The students will perform a written exam similar to June in which they will obtain a maximum of 5 points

# Sources of information Basic Bibliography Complementary Bibliography A. Delgado C. Minguillón y J. Juglar, Introducción a la Química Terapéutica, 2ª Edición 2003, G. L. Patrick, An introduction to Medicinal Chemistry, 6th Edition 2017, C. G. Wermuth, 4. The Practice of Medicinal Chemistry, 4th Edition 2015, R. Renneberg, Biotecnología para principiantes, 2004,

### Subjects that it is recommended to have taken before

Biology: Biology/V11G200V01101

IT tools and communication in chemistry/V11G200V01401

Physical chemistry I/V11G200V01303

Physical chemistry II/V11G200V01403

Organic chemistry I/V11G200V01304

Structural Determination/V11G200V01501

Chemical engineering/V11G200V01502

Analytical chemistry II/V11G200V01503

Biological chemistry/V11G200V01602

Organic chemistry II/V11G200V01504

Organic chemistry III/V11G200V01704