# $Universida_{\hbox{\it de}}\!Vigo$

Subject Guide 2015 / 2016

IDENTIFYIN	<u> </u>			
	communication in chemistry			
Subject	IT tools and			
	communication in			
-	chemistry			
Code	V11G200V01401			
Study	(*)Grao en Química			
programme				
Descriptors	ECTS Credits	Choose	Year	Quadmester
	6	Mandatory	2nd	2nd
Teaching	English			
language		1	,	
Department				
Coordinator	Correa Duarte, Miguel Ángel			
Lecturers	Correa Duarte, Miguel Ángel			
	Pérez Juste, Jorge			
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Web				
General	The course aims to familiarize students with the	use of chemical inform	nation sources (	scientifical and technical
description	in general) with emphasis on its use through the	Internet, as well as wi	th the use of al	l types of software tools
	for statistical calculations and chemical modeling . Attention is also paid to the acquisition of important			
	communication skills (writing scientific and technical documents, academic, web design, etc).			

Com	petencies
Code	
C22	Process and perform computational calculations with chemical information and chemical data
C23	Present oral and written scientific material and scientific arguments to a specialized audience
D1	Communicate orally and in writing in at least one of the official languages of the University
D2	Communicate at a basic level in English in the field of chemistry
D3	Learn independently
D4	Search and manage information from different sources
D5	Use information and communication technologies and manage basic computer tools
D6	Use mathematics, including error analysis, estimates of orders of magnitude, correct use of units and data representations
D7	Apply theoretical knowledge in practice
D8	Teamwork
D9	Work independently
D10	Work at a national and international context
D14	Analyze and synthesize information and draw conclusions
D15	Evaluate critically and constructively the environment and oneself
D16	Develop an ethical commitment
D18	Generate new ideas and show initiative

Learning outcomes		
Expected results from this subject	Training and Learning Results	
To know the different sources of scientific and technical information	C23 D1 D2 D4 D5 D9 D14 D16	

To understand the basics of running a Science lil of its services	brary and know how to perform an advanced use		D2 D4 D5 D8 D9 D14
To classify scientific journals based on their then	ne or objective	C23	D1 D2 D3 D5 D8 D9 D10 D15
To know the basic characteristics of other source patents, dissertations, government publications, directories, databases and """handbooks""".		C23	D1 D2 D5 D8 D10 D16
To know the basic characteristics of other sources: technical reports, conference proceedings, patents, dissertations, government publications, standards, videos, dictionaries, encyclopedias, directories, databases and """"handbooks"""".			D1 D2 D5 D8 D10 D16
To know the structure and function of an abstract	cting or indexing service	C23	D1 D2 D5 D8 D10 D16
To know how to use statistical program package of statistical analysis	s to perform data fitting, graphical and other kind	s C22	D3 D5 D6 D7 D9 D14 D16
Contents			
Topic			
The scietific literature: general aspects.	Structure and classification of the literature.		·
	General rules of a literature search.  Function, organization and use of a scientific lib	rarv.	
Information Sources	Books. Journals. Technical reports. Conference Proceedings. Patents. Thesis. Government Publications. Standards. Videos. Dictionaries. Directories Encyclopedias Databases		

Using Internet	Basic Internet services.
	Remote connection and file transfer utilities.
	Search engines.
	Electronic lists and subscription services.
	Other services.
Indexing and abstracting services	Structure, function and design of web pages.  Identification of a scientific paper.
indexing and abstracting services	The ISI Web of Knowledge (WOK).
	The Chemical Abstract Service (CAS) and the Scifinder.
	Other abstracting services.
	Handbooks.
Bibliographic Managers	Classification of bibliographic references: general principles.
	Use of popular software packages:
	Refworks and Endnote as examples.
Preparation of a scientific, technical or academic document	
	References, tables and figures : general principles.
	Use of computer templates.
	General aspects of the scientific style and the use of English.
	How to write: CVs, progress reports, grant requests and other academic documents.

Class hours	Hours outside the classroom	Total hours
14	28	42
26	52	78
2	22	24
1.5	4.5	6
	Class hours  14 26 2 1.5	classroom

<sup>\*</sup>The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

	Description
Master Session	The theoretical aspects of the subject are presented
Practice in computer	Computer lab exercises: literature searches, use of bibliographic managers, use of statistical
rooms	packages, report writing.
Troubleshooting and /	or Report or article writing in English language.
exercises	Simple exercises with modelling software

Personalized attention			
Methodologies	Description		
Practice in computer rooms	The student is helped by providing adequate guidelines. Since all lectures are given in the computer room, the student will be helped mostly there in a practical and effective way.		
Troubleshooting and / or exercises	The student is helped by providing adequate guidelines. Since all lectures are given in the computer room, the student will be helped mostly there in a practical and effective way.		

Assessment		
	Description	Qualification Training and Learning
		Results

Practice in computer rooms	Typically, literature searches	20	C22 C23	D1 D2 D3 D4 D5 D9 D15 D16
Troubleshooting and / or exercises	Tipically, database searches and use of utilities of modelling software.	40	C22 C23	D1 D2 D3 D4 D5 D8 D10 D14 D15 D18
Long answer tests and developmen	ntWritten exam consisting of short questions.	40	-	D1 D2 D14 D15

#### Other comments on the Evaluation

Attendance at practical lectures (seminars) is compulsory. The student will be given a rating (0-10) as long as he/she has attended 3 or more seminar sessions, has delivered at least two reports on the exercises or practices proposed by the teacher or has done a written exam.

If the student fails in the first call he/she will be asked to improve some of the exercises or perform new ones provided by the teacher. In addition he/she will have to undergo a more thorough exam, which will weight 50% of the final grade.

#### **Sources of information**

Douville, J.A., The literature of chemistry, 1st,

Kaplan, S.M., The English-Spanish Spanish-English dictionary of chemistry, 2ª,

Day, R.A.; Gastel, B., How to write and publish a scientific paper, 7ª,

## Recommendations

### Subjects that are recommended to be taken simultaneously

Numerical methods in chemistry/V11G200V01402 Physical chemistry II/V11G200V01403

Inorganic chemistry I/V11G200V01404

### Subjects that it is recommended to have taken before

Physics: Physics I/V11G200V01102
Physics: Physics II/V11G200V01201
Chemistry: Chemistry I/V11G200V01105
Chemistry: Chemistry II/V11G200V01204