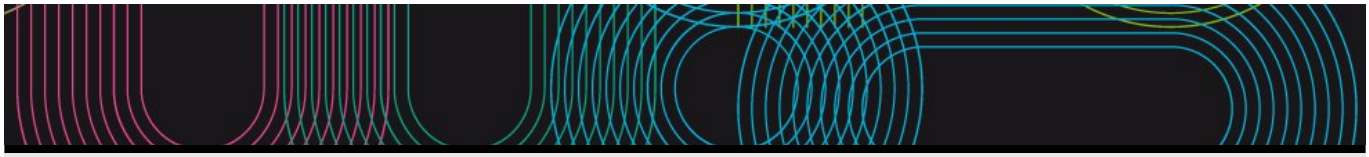

IDENTIFYING DATA				
Thermal engineering I				
Subject	Thermal engineering I			
Code	V12G770V01401			
Study programme	PCEO Grado en Ingeniería Mecánica/Grado en Ingeniería en Electrónica Industrial y Automática			
Descriptors	ECTS Credits	Choose	Year	Quadmester
	9	Mandatory	4th	1st
Teaching language	Spanish Galician			
Department				
Coordinator	Cerdeira Pérez, Fernando			
Lecturers	Araújo Fernández, Enrique José Cerdeira Pérez, Fernando Diz Montero, Rubén Pequeño Aboy, Horacio			
E-mail	nano@uvigo.es			
Web	http://moovi.uvigo.gal/			
General description	Acquisition of knowledges to comprise the operation of the thermal machines and the processes that take place in his interior, as well as know the types of machines and installations more important and his components. His knowledge results basic for the analysis of the operation, design and construction of the thermal machines and of the thermal teams associated to the same, and in general the industrial applications of the thermal engineering.			

Training and Learning Results
Code

Expected results from this subject	
Expected results from this subject	Training and Learning Results

Contents	
Topic	
Foundations of the combustion.	Introduction. Types of combustion.
Fuels employed in engines and thermal installations.	Classification. Properties. Distribution of gases fuels
Study of the humid air.	Introduction. Variables psycrometrics. Diagrams psycrometric. Cooling tower.
Heat Exchangers.	Introduction. Classification Thermal balance. Distribution of temperature Analysis of exchangers - Method DTLM - Method NTU
Machines and thermal engines.	Classification. Basic concepts.

Engines of internal combustion.	Real cycles and theorists. Main components. Parameters characteristics. Characteristic curves. Auxiliary systems: refrigeration and lubrication.
Installations of power with cycle of steam.	Introduction. Main components. Cycle Rankine. Thermal balance.
Installations of power with cycles of gas.	Introduction. Main components. Cycle Brayton. Thermal balance. Cycle Combined of gas-steam.
Pumping of heat.	Definitions. Cycle of Carnot reverse. Cycle of mechanical compression. Bomb of heat. Refrigeration by absorption. Refrigerants.
Boilers and Burners.	Classification. Definitions. Types. Energetic balance.
Compressors.	Previous concepts. Reciprocating compressors. Rotary compressors.
Processes of spill.	Properties of stagnation. Speed of the sound and nº of Mach. Flow isentropic through nozzles and diffusers.
-- Laboratory practices.	- Determination of the enthalpy of combustion. - Study of the flame propagation. - Higrometric study of the air. - Study of the heat exchangers. - Study of the engines of 2T. - Study of the engines of 4T. - Study of the air compressors. - Energetic balance of a boiler. - Visit to a boilers room.
-- Practical with support of the TIC	- Calculation of a LPG deposit.

Planning

	Class hours	Hours outside the classroom	Total hours
Lecturing	38	32	70
Problem solving	14	32	46
Practices through ICT	4	2	6
Laboratory practical	18	12	30
Mentored work	0	4	4
Autonomous problem solving	0	25	25
Field practice	2	2	4
Problem and/or exercise solving	3	12	15
Objective questions exam	1	4	5
Objective questions exam	1	4	5
Objective questions exam	1	4	5
Objective questions exam	2	8	10

*The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

Methodologies

	Description
Lecturing	Exhibition by part of the professor of the contents of the matter object of study.
Problem solving	Resolution of problems and/or exercises related with the subject that the student will make in classroom and/or laboratory. They will resolve problems of character "type" and/or practical examples.
Practices through ICT	Simulation of processes related with the content of the matter using specific software.
Laboratory practical	Experimentation of real processes in laboratory that complement the contents of the matter.

Mentored work	Activity directed to develop exercises or projects under the guidelines and supervision of the professor. His development can be linked with autonomous activities of the student, practices of laboratory,.... Activity in group or individual. The work developed can finally be exposed publicly in the classroom.
Autonomous problem solving	Resolution of problems and/or exercises related with the subject that the student will make out of the classroom.
Field practice	A visit to a boiler room is made to learn safety aspects of engineering practice. Students will be expected to identify potential risks, existing protection measures or devices or safety distances, among others, and when possible, measures will be taken to learn about both energy efficiency and environmental aspects. To do this, it asks them to carry out a prior bibliographic search of the mandatory standard. This action is completed with a brief questionnaire on notions of industrial health and safety.

Personalized assistance

Methodologies	Description
Autonomous problem solving	The students will be able to resolve the doubts of the matter and of the distinct bulletins of problems in the schedule of tutorials fixed by the professors of the matter.

Assessment

	Description	Qualification	Training and Learning Results
Laboratory practical	Presentation of a memory, podcast or similar in which the practices developed in the laboratory are described.	5	
Problem and/or exercise solving	Final exam of problems or practical cases. It will be mandatory to obtain a minimum score of 3.5 out of 10 in this test.	40	
Objective questions exam	Objective test (1) consisting of a problem or short questions and/or test type to know the progressive evolution of the students during the development of the subject.	10	
Objective questions exam	Objective test (2) consisting of a problem or short questions and/or test type to know the progressive evolution of the students during the development of the subject.	10	
Objective questions exam	Objective test (3) consisting of a problem or short questions and/or test type to know the progressive evolution of the students during the development of the subject.	10	
Objective questions exam	Objective test (4) consisting of a problem or short questions and/or test type to know the progressive evolution of the students during the development of the subject. It will be mandatory to obtain a minimum score of 3.5 out of 10 in this test.	25	

Other comments on the Evaluation

On the second opportunity (July session), students who have chosen the continuous assessment (CA) modality may choose, prior to taking the exam (> 24 h), between keeping the CA mark or taking a specific test (ST).
Both the students who have chosen the modality of global evaluation according to the procedure and the term established by the school and those who go to the End of Degree call will be evaluated by means of a global exam (100%) made up of theory and problems.

A numerical rating system of 0 to 10 points will be used according to current legislation (RD 1125/2003, September 5, BOE September 18).

Ethic Commitment: It is expected an adequate ethical behaviour of thestudent. In case of detecting unethical behaviour (copying, plagiarism,unauthorized use of electronic devices, etc.) shall be deemed that the studentdoes not meet the requirements for passing the subject. In this case, theoverall rating in the current academic year will be Fail (0.0).
The use of any electronic device for the assessmenttests is not allowed unless explicitly authorized. The fact of introducing unauthorized electronic device in the examination room will be considered reason for not passing the subject in the current academic year and will hold overall rating (0.0).

Sources of information

Basic Bibliography

Agüera Soriano, José, **Termodinámica lógica y motores térmicos**, Ciencia 3, D.L., 1999
Moran M.J.; Shapiro H.N., **Fundamentos de termodinámica técnica**, 2ª/4ª, Editorial reverté, S.A., 2004
Çengel Y.A.; Boles M.A., **Termodinámica**, 6ª, McGraw-Hill-Interamericana, 2009

Incropera, Frank P., **Fundamentos de transferencia de calor**, 4ª, Prentice Hall, 1996

Complementary Bibliography

Potter M.C.; Somerton C.W., **Termodinámica para ingenieros**, 1ª, McGraw-Hill/Interamericana de España, D.L., 2004
Múñoz Domínguez, M.; Rovira de Antonio, A.J., **Ingeniería Térmica**, UNED, 2006
Çengel Y.A.; Ghajar, A.J., **Transferencia de calor y masa**, 4ª, McGraw-Hill/Interamericana de España, D.L., 2011
Kohan, Anthony L., **Manual de calderas**, 4ª, McGraw-Hill, 2000

Recommendations

Subjects that it is recommended to have taken before

Physics: Physics I/V12G380V01102
Physics: Physics II/V12G380V01202
Chemistry: Chemistry/V12G380V01205
Thermodynamics and heat transfer/V12G380V01302

Other comments

To enrol in this subject is necessary to have surpassed or be enrolled of all the subjects of the inferior courses.