Universida_{de}Vigo

Subject Guide 2018 / 2019

IDENTIFYIN	G DATA	TO WALLET	,	7,7,7,7,7,7
Environmen	ital chemistry			
Subject	Environmental			
	chemistry			
Code	V11G200V01902			
Study	(*)Grao en			
programme	Química			
Descriptors	ECTS Credits	Choose	Year	Quadmester
	6	Optional	4th	2nd
Teaching	Spanish			
language	English			
Department	Analytical and Food Chemistry		-	
•	Physical Chemistry			
Coordinator	González Romero, Elisa			
Lecturers	González Romero, Elisa			
	Pérez Juste, Jorge			
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Web				
General description	Global knowledge of the chemic quality, treatment and manage			

Competencies

Code

- C2 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: types of chemical reactions and its main characteristics
- C4 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: Basics and tools for solving analytical problems and characterization of chemical substances
- C17 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories in: metrology of chemical processes including quality management
- C18 Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: principles of electrochemistry
- D1 Communicate orally and in writing in at least one of the official languages of the University
- D3 Learn independently
- D4 Search and manage information from different sources
- D5 Use information and communication technologies and manage basic computer tools
- Use mathematics, including error analysis, estimates of orders of magnitude, correct use of units and data representations
- D7 Apply theoretical knowledge in practice
- D8 Teamwork
- D9 Work independently
- D10 Work at a national and international context
- D12 Plan and manage time properly
- D13 Make decisions
- D14 Analyze and synthesize information and draw conclusions
- D15 Evaluate critically and constructively the environment and oneself
- D16 Develop an ethical commitment
- D17 Develop concern for environmental aspects and quality management

Expected results from this subject

Training and Learning
Results

Describe the main chemical processes that occur in each layer of the atmosphere. Describe the mechanisms of production and destruction of ozone. Explain the greenhouse effect	C2 C17	D1 D3 D4 D5 D6 D7 D8 D9 D10 D12 D13 D14 D15 D16 D17
Describe the composition and properties of the natural waters	C2 C17	D1 D3 D4 D5 D6 D7 D8 D9 D10 D12 D13 D14 D15 D16 D17
Explain the exchange of matter between the distinct environmental compartments. Time of residence	C2 C17	D1 D3 D4 D5 D6 D7 D8 D9 D10 D12 D13 D14 D15 D16 D17
Explain the main causes of the corrosion and how minimise it	C2 C18	D3 D4 D5 D6 D7 D9 D10 D14 D16 D17
Identify the main pollutants present in the natural media and the main pollutants according to the different environmental rules	C2 C4 C17	D3 D4 D5 D6 D7 D9 D10 D13 D14 D16 D17

Recognise the different types of chemical reactions that experience the pollutants in the natural medias	C2 C4 C17	D3 D4 D5 D6 D7 D10 D14
Estimate the harmful effects for the environment of the diverse types of pollutants	C2 C4 C17	D17 D3 D4 D5 D6 D7 D8 D9 D10 D13 D14 D16 D17
Describe the sampling, pre-treatment and preparation of sample for the analysis of environmental pollutants	C4 C17	D3 D4 D5 D6 D7 D8 D10 D13 D14 D16 D17
Select the appropriate analytical techniques and the concrete methods for its determination in the atmosphere, waters, floors, sediments and biota	C4 C17	D3 D4 D5 D6 D7 D8 D10 D13 D14 D15 D16 D17
Describe the main available technologies for the treatment of the pollution and evaluate its applicability in diverse cases	C4	D1 D4 D5 D6 D7 D8 D10 D12 D13 D14 D15 D16 D17

 C4 C17	D1 D4 D5 D6 D7 D8 D10 D12 D13 D14
	D14
	D15
	D16
	D17

Contents	
Topic	
	Comparalista
1 The matter and its cycles	Generalities
2 Chemical processes in the atmosphere	Photochemical processes. Chemistry of
	the layer of ozone. Greenhouse effect .
3 Chemical processes in the hydrosphere	Salinity and alkalinity. Transfer of
	matter between environmental compartments. Interface Atmosphere-
	water. Exchange of gases. Interface Sediment-water
4 Electrochemical processes in the environme	nt Corrosion
5 Environmental Pollutants	Classification. Natural transformations
	of the pollutants.
6 Analysis of pollutants	Analytical methodology: sampling and treatment of sample, techniques
	and methods in the determination of pollutants. Applications in
	atmosphere, waters, floors, sediments and biota
7 Quality Control in the laboratories of	Generalities
environmental analysis	
8 Quality Assurance of the pollution	Generalities
9 Evaluation of the environmental impact	Systems of environmental management

	Class hours	Hours outside the	Total hours
	Class floars	classroom	Total Hours
Seminars	10	25	35
Presentation	4	14	18
Scientific events	3	4.5	7.5
Workshops	0	12	12
Lecturing	22	33	55
Short answer tests	2	9	11
Essay questions exam	2	9.5	11.5

^{*}The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

Methodologies	
	Description
Seminars	The aim that pursues in the seminars is to settle the knowledges and expand the competitions purchased in the masterclasses, giving practical and representative examples of the fundamental concepts that collect in each subject.
Presentation	Each student will choose, to the start of the course, a subject of which suggest, or another if it is of interest for him, but always related with the program of the Environmental Chemical matter, and will realise a diagram and synthesis of the work to be exposed in a maximum time of 10 min, in which it will include a practical example extracted of one or several scientific articles. The aims to cover are: introduction and/or practical in the bibliographic research, preparation and presentation of the scientific work, comparison of results between different technical, evaluation of the environmental impact, etc Previous to the exhibition, the student/to will deliver, in a dossier with his name and title of the exhibition, a copy of all the articles consulted and of the presentation of the same. The assistance to the exhibitions is compulsory and any of the questions formulated during his development can fall in the examinations

Scientific events	They include other less conventional activities inside the program of the matter, like the assistance to conferences, webinars of the ACS, "workshops" or congresses that celebrate in the own University, what will allow to the student expand his horizons and begin to go in in contact with other realities further of the faculty, obtaining information at first hand through representatives of companies, of professors of other universities -and, even, of other countries - that will orient them on other opportunities and will promote the mobility of these students. Of this form, pretends transmit to the student the multiple possibilities that can him present in the future, showing him a fan of labour possibilities. These events are subject to the programmings extra-academic of the different centres in the own University, but in any moment overlap with activities programmed previously and, in his case, would look for other alternatives.
Workshops	They would form part of the seminars in which the students will have to resolve by himself same, under the supervision of the professor but with a greater autonomy, real practical suppositions of chemical processes, detection of possible pollutants in which they derive, the environmental impact that produce and design strategies for his control
Lecturing	The masterclasses (55 min) pretend to give a global and real vision of the chemical processes that produce in the environment, the interaction between the different compartmentalized means, the pollutants present and those that generate, the most appropriate methodology for his analysis and his environmental control. Each one of the subjects will go documented with scientific articles, whose contents will serve to settle and expand the knowledges purchased in the theoretical classes, and of representative examples of the fundamental concepts that collect each subject. The methodology education-learning will be centred in the student, by what the classes will be headed to motivate a high participation by part of these in the classroom. The platform *Tem@ will be the resource that allow to the student the communication with the professor and his mates, through a virtual application, at the same time to be the source of information of immediate access for them. In her they will be able to find the basic information and documentation on the matter that gives, the diary of activities, the exercises to realise and the qualifications.

Methodologi	ies Description
Seminars	In the seminars and in the workshops will do a follow-up of the personal work that was realising the student in this moment, related with the matter. They realised experiments of classroom, useful for the problems resolution, including the oral exposition and other complementary works that propose, in function of the evolution of the student in the process of learning
Workshops	In the seminars and in the workshops will do a follow-up of the personal work that was realising the student in this moment, related with the matter. They realised experiments of classroom, useful for the problems resolution, including the oral exposition and other complementary works that propose, in function of the evolution of the student in the process of learning

Assessment				
	Description	Qualification	Lea	ng and rning sults
Presentation	The presentations and other activities associated (ACS Webinars, conferences and Meeting/Symposiums) until arriving to the defence of the work.	20	C17	D1 D3 D4 D5 D8 D9 D10 D14 D16
Short answer tests	They will realise two short proofs of one or two hours of length, C1 and C2, along the quatrimester in which it gives the matter and whose dates will be fixed in the chronogram to the start of the course. They are not eliminatory.	30	C2 C4 C18	D1 D3 D6 D7 D12 D13 D14 D15 D16

Essay questions	The long proof (divided into 2 parts) will have until three hours and in her will	50	C2	D1
exam	go in all the subjects given of the matter and the activities associated to them.		C4	D3
	A minimum of 4 in each part is required to be compensated by both parties		C18	D6
				D7
				D12
				D13
				D14
				D15
				D16

Other comments on the Evaluation

All the partial qualifications will allow to make the final qualification, valuing the attitude of participation and the interest showed by the student along the course. Due to the fact that each one of the subjects will go documented with scientific articles, some question extracted of them will be able to form part of the short proofs and/or long and in the second announcement.

It considers no-presented (NP) not assisting to 25% of the face-to-face hours and/or not realising any of the proofs (short or long) neither participate in the activities programmed. In the moment in that any of the parts have qualification, in records will appear said qualification obtained, although it have not realised any another proof or activity programmed.

In the second announcement, the students will have the opportunity to recover 50% of the matter. This proof contemplates the same contents that require for the long proof and will keep the qualifications of the others sections evaluated along the course.

To achieve approve the matter, the students will have to surpass 50% of all and each one of the proofs and activities program of the matter.

Sources of information

Basic Bibliography

Complementary Bibliography

P.W. ATKINS, Química Física,

I.N. LEVINE, Fisicoquímica,

Stanley E. Manahan, Environmental Chemistry, 9,

Roger N. Reeve, Introduction to Environmental Analysis,

F. W. Fifield y P. J. Haines (Editores), Environmental Analytical Chemistry, 2,

Frank M. Dunnivant, Environmental Laboratory Exercises for Instrumental Analysis and Environmental Chemistry, Chunlong Zhang, Fundamentals of Environmental Sampling and Analysis,

J. P. RILEY y G. SKIRROW, Chemical Oceanography,

ISI WEB OF KNOWLEDGE.

Scifinder,

Environmental Sciences Category,

Colin Baird y Michael Cann, QUIMICA AMBIENTAL, 2ª edición,

Recommendations

Subjects that continue the syllabus

Final Year Dissertation/V11G200V01991

Subjects that are recommended to be taken simultaneously

Industrial chemistry/V11G200V01904

Final Year Dissertation/V11G200V01991

Subjects that it is recommended to have taken before

Analytical chemistry 1/V11G200V01302

Physical chemistry I/V11G200V01303

Physical chemistry II/V11G200V01403

Analytical chemistry II/V11G200V01503

Analytical chemistry 3/V11G200V01601

Physical chemistry III/V11G200V01603