



IDENTIFYING DATA

Analytical chemistry 3

Subject	Analytical chemistry 3			
Code	V11G200V01601			
Study programme	(*)Grao en Química			
Descriptors	ECTS Credits	Choose	Year	Quadmester
	6	Mandatory	3rd	2nd
Teaching language	Spanish			
Department	Analytical and Food Chemistry			
Coordinator	Bendicho Hernández, José Carlos			
Lecturers	Bendicho Hernández, José Carlos Lavilla Beltrán, María Isela			
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General description	<p>"Machine translation into english of the original teaching guide" - This matter provides to the students the knowledge on important and actual aspects on Analytical Chemistry (Chemometrics; Trace Analysis; Automatism and sensors), especially those regarding strategies that have allowed the evolution of the conventional methodologies to improve the quality of the analytical information. Students will be able to complement his training by means of the integration of the knowledge of Analytical Chemistry taken previously, specially the contents in Analytical Chemical II (introduction to the instrumental analysis). This will allow them to tackle the resolution of analytical problems in different areas of interest (environment, feeding, industry, clinic etc.).</p>			

Competencies

Code	
A1	Students have demonstrated knowledge and understanding in a field of study that builds upon their general secondary education, and is typically at a level that, whilst supported by advanced textbooks, includes some aspects that will be informed by knowledge of the forefront of their field of study
A2	Students can apply their knowledge and understanding in a manner that indicates a professional approach to their work or vocation, and have competences typically demonstrated through devising and sustaining arguments and solving problems within their field of study
A3	Students have the ability to gather and interpret relevant data (usually within their field of study) to inform judgments that include reflection on relevant social, scientific or ethical issues
C4	Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: Basics and tools for solving analytical problems and characterization of chemical substances
C8	Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: main techniques for structural determination, including spectroscopy
C17	Demonstrate knowledge and understanding of essential facts, concepts, principles and theories in: metrology of chemical processes including quality management
C18	Demonstrate knowledge and understanding of essential facts, concepts, principles and theories: principles of electrochemistry
C19	Apply knowledge and understanding to solve basic problems of quantitative and qualitative nature
C20	Evaluate, interpret and synthesize data and chemical information
C22	Process and perform computational calculations with chemical information and chemical data
C24	Recognize and analyze new problems and plan strategies to solve them
C29	Demonstrate skills for numerical calculations and interpretation of experimental data, with special emphasis on precision and accuracy
D1	Communicate orally and in writing in at least one of the official languages of the University
D3	Learn independently
D4	Search and manage information from different sources
D5	Use information and communication technologies and manage basic computer tools
D6	Use mathematics, including error analysis, estimates of orders of magnitude, correct use of units and data representations

D7 Apply theoretical knowledge in practice

D8 Teamwork

D9 Work independently

D12 Plan and manage time properly

D13 Make decisions

D14 Analyze and synthesize information and draw conclusions

D17 Develop concern for environmental aspects and quality management

Learning outcomes

Expected results from this subject	Training and Learning Results		
1. Select and apply distinct technical *quimiométricas to the resolution of practical cases and justify the utilisation of the same.	A1	C17	D1
	A2	C19	D3
	A3	C20	D5
		C22	D6
			D7
			D9
			D13
			D14
			D17
2. Use the experimental design like tool for the optimisation of an analytical method.	A1	C17	D1
		C19	D3
		C22	D5
			D6
			D7
			D9
			D13
			D14
4. Justify the utilisation of the Chemometrics in the quality of the results. Describe how implements a system of quality in a laboratory of control of analytical.	A1	C4	D1
	A2	C17	D3
		C19	D5
		C20	D6
		C29	D7
			D8
			D9
			D14
			D17
3. Evaluate and interpret the analytical results of systems *multicomponentes and *multivariables.	A1	C4	D1
	A2	C17	D3
	A3	C20	D5
		C22	D6
			D7
			D8
			D9
			D13
			D17
6. Recognise the different methods of treatment of sample as well as evaluate his possibilities in the resolution of diverse analytical problems inside the field of the analysis of trace.	A1	C4	D1
	A2	C19	D3
		C20	D4
			D7
			D9
			D12
			D13
			D14
			D17
5. Describe the planning of the sampling and the factors that take part in him for the analysis of trace.	A1	C4	D1
		C17	D3
		C24	D4
			D6
			D7
			D9
			D12
			D13
			D17

7. Compare and value the different methods of existent extraction in the actuality, like the extraction by fluent *supercríticos, in solid phase, *microextracción, etc.	A1 A2	C4 C19 C20	D1 D3 D8 D9 D12 D14 D17
8. Describe the analytical methodology and instrumentation as well as know the applications of technicians of general use in analysis of trace like the voltammetry of *redisolución *anódica, spectrometry of atomic absorption with atomisation *electrotérmica, spectrometry of masses with source of plasma and the different attachments between the chromatography and the spectrometry of masses.	A1	C4 C8 C18 C19	D1 D3 D4 D8 D9
9. Classify the different types of automatic systems and *miniaturizados, establishing his advantages and inconvenient, modalities and applications more notable and of immediate future. Justify the automation in the different stages of the analytical process.	A1 A2	C4 C17 C20	D1 D3 D4 D5 D8 D9 D17
10. Explain the foundations of the sensors and *biosensores chemical, as well as his more important applications. Explain and value the importance of the utilisation of the sensors for the fast and reliable obtaining of analytical information.	A1 A2 A3	C4 C17 C20	D1 D3 D4 D8 D9 D12
11. Describe the characteristics of the continuous automatic analysers, discontinuous and *robotizados. Know the phenomena of dispersion in continuous analysers of injection in flow and of sequential injection, as well as the form to characterise them.	A1	C4 C17 C19 C20	D1 D3 D4 D5 D8 D9 D14 D17
12. Explain the construction of analytical tools in miniature and his applications.	A1	C4 C17 C19	D1 D3 D4 D5 D9 D12 D14

Contents

Topic

SUBJECT 1. Analysis of trace	Concept and importance of the analysis of trace. Sources of pollution in the laboratory. Experimental methods in analysis of trace. Sampling. Methods of decomposition in analysis of trace inorganic. Methods of extraction in analysis of trace organic. Technicians selected of analysis of trace.
SUBJECT 2. Automation	Automation in the laboratory of analysis: generalities. Automatic analysers. Discontinuous analysers, continuous and *robotizados. Analysers of injection in flow and flow *segmentado: characteristics. Phenomena of dispersion. Characteristics of the signal of injection in flow. Technicians of gradient. Analysers of sequential injection. Instrumentation and applications.
SUBJECT 3. Sensors and *biosensores chemical	Concept of sensor. Components of a chemical sensor. Classification. Sensors and *biosensores. Elements of recognition. Types of *transductores. (*Bio)Electrochemical and optical sensors. Applications of interest. Miniaturisation of analytical systems.
SUBJECT 4. Introduction to the Chemometrics	Definition and historical evolution of the Chemometrics. The chemometrics in the different stages of the analytical process. Basic statistical concepts. Parameters that estimate the central value and the dispersion: parametric and no parametric. Properties of the variance and the average. Expression of analytical results.

SUBJECT 5. Basic chemometrics: comparison of analytical results	Test of significance. Proofs of hypothesis: structure of the proofs of hypothesis. Errors type I and II. Probability. Rejection of anomalous results. Parametric proofs of comparison of two variances. Parametric proofs of comparison of two averages. Comparison of several half *muestrales by means of *ANOVA of a road. Control of the accuracy and precision over time: charts of control. Proofs no parametric.
SUBJECT 6. The quality in the analytical laboratories: *cualimetría.	Introduction to the *cualimetría: quality and chemometrics. Quality and analytical properties: validation of analytical methods. *Trazabilidad. Generic approximation to the quality. Systems of quality: Norms ISO. Accreditation and certification of the laboratories.

Planning

	Class hours	Hours outside the classroom	Total hours
Seminars	13	26	39
Supervised work	0	9	9
Lecturing	26	52	78
Short answer tests	2	4	6
Short answer tests	2	4	6
Essay questions exam	4	8	12

*The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

Methodologies

	Description
Seminars	In the classes of seminar will reinforce the learning of the *temario explained during the sessions *magistrales, carrying out the resolution of numerical problems and theoretical exercises-practical. The professor will propose, of regular form, different problems/exercises that will be resolved of individual form by the student and delivered for his evaluation.
Supervised work	It will provide to the student a series of articles published in magazines of education in Chemistry and related with the contents of the matter. Once studied the article, the student will have to answer to a questionnaire of questions provided by the professor.
Lecturing	The professor will develop the contents of the program from the proportionate material to the student through the platform FEAR. In the sessions *magistrales, the professor will present the fundamental appearances of the matter that will have to complement by means of the bibliography recommended.

Personalized attention

Methodologies	Description
Lecturing	The professor will resolve the doubts of personalised way on any one of the activities proposed (masterclasses, seminars, works *tutelados, resolution of problems/exercises and proofs). To such end, the professor will inform the available schedule in the presentation of the matter.
Seminars	The professor will resolve the doubts of personalised way on any one of the activities proposed (masterclasses, seminars, works *tutelados, resolution of problems/exercises and proofs). To such end, the professor will inform the available schedule in the presentation of the matter.

Assessment

	Description	Qualification	Training and Learning Results
Seminars	In the classes of seminar, the professor will resolve part of the problems/exercises, leaving others to be resolved by the student. The delivery of the problems/exercises resolved is compulsory. To be able to evaluate is activity, the student will have to carry out at least 75% of the deliveries. Besides it will be necessary to obtain a minimum punctuation of 3 on 10 points so that the qualification of this activity can add to the rest of elements of evaluation.	10	A1 C4 D6 A2 C8 D7 A3 C17 D9 C18 D12 C19 D14 C20 C22
Supervised work	The realisation of the works is compulsory. So that this activity can be evaluated, the student will have to carry out at least 75% of the deliveries. Besides it will be necessary to obtain a minimum punctuation of 3 on 10 points so that the qualification of this activity can add to the rest of elements of evaluation.	5	A1 C4 D1 A2 C8 D3 A3 C17 D4 C18 D5 C19 D7 C20 D8 C24 D9 D14 D17

Short answer tests	It will effect a first short proof on the subjects 1, 2 and 3, roughly to half of the *cuatrimestre. The short proof will be able to consist in questions of short answer, problems and ask type test. The presentation to this proof *inhabilita to the student to obtain the qualification of no presented.	20	A1 A2 A3	C4 C8 C17	D1 D6 D7 D9 D12 D13 D14
Short answer tests	It will effect a second short proof on the subjects 4, 5 and 6 to the end of the *cuatrimestre. The short proof will be able to consist in questions, problems and exercises. The presentation to this proof *inhabilita to the student to obtain the qualification of no presented.	25	A1 A2 A3	C4 C17 C19	D1 D6 D7 D9 D12 D13 D14
Essay questions exam	Compulsory final examination. It will consist in a global proof of the *temario that will include problems, exercises and ask type test. It will be necessary to obtain 3 points on 10 in this examination so that the qualification can add to the one of the rest of elements of evaluation.	40	A1 A2 A3	C4 C8 C17	D1 D6 D7 D9 D12 D13 D14 C24

Other comments on the Evaluation

The participation of the student in any one of the activities evaluated (deliveries of problems and exercises, proofs of short answer) *inhabilita to the student to obtain the qualification of NO PRESENTED. To surpass the short proofs as well as the long proof (final examination), will be necessary that exist a balance in the qualifications of the theoretical part and of the problems.

ANNOUNCEMENT OF JULIO: The qualification in this announcement will be formed by two components: 1. Punctuations obtained by the student during the course (maximum 5 points) They will keep the qualifications in the works *tutelados (maximum 0.5 points), problems/exercises resolved (maximum 1 point) and short proofs (maximum 3.5 points). 2. Global written proof of the contents of the matter (maximum 5 points) This proof will include problems, exercises and ask type test. To be able to approve in this announcement, the student has to obtain at least 3 points on 10 in this proof. The presentation to this proof *inhabilita to the student to obtain the qualification of NO presented.

Sources of information

Basic Bibliography

G. Ramis Ramos; M.C. Álvarez Coque, **Quimiometría**, Síntesis, 2001
 J.C. Miller; J.N. Miller, **Estadística y Quimiometría para Química Analítica**, Prentice-Hall, 2002
 R. Compañó Beltrán; R. Ríos Castro, **Garantía de calidad en los laboratorios analíticos**, Síntesis, 2002
 C. Cámara, **Toma y tratamiento de muestras**, Síntesis, 2002
 R. Cela, **Técnicas de separación en Química Analítica**, Síntesis, 2002
 C. Cámara, **Análisis químico de trazas**, Síntesis, 2011
 Valcárcel, **Automatización y miniaturización en Química Analítica**, Springer, 2000

Complementary Bibliography

S. Mitra, **Sample preparation techniques in analytical chemistry**, Wiley, 2003
 B.R. Eggins, **Chemical sensors and biosensors**, Wiley, 2002
 L. Hernández, **Introducción al análisis instrumental**, Ariel, 2002
 K.A. Rubinson, **Análisis Instrumental**, Prentice-Hall, 2000
 Skoog, **Principios de Análisis Instrumental**, McGraw-Hill, 2001
 Kellner, **Analytical Chemistry**, Wiley-VCH, 2004
 M. Valcárcel, M.D. Luque de Castro, **Flow-injection analysis. Principles and applications**, Ellis Horwood, 1987

Recommendations

Subjects that it is recommended to have taken before

Analytical chemistry I/V11G200V01302
 Analytical chemistry II/V11G200V01503