# Universida<sub>de</sub>Vigo

Subject Guide 2017 / 2018

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IT tools and			
communication in			
chemistry			
V11G200V01401			
(*)Grao en Química			
ECTS Credits	Choose	Year	Quadmester
6	Mandatory	2nd	2nd
English			
Correa Duarte, Miguel Ángel			
Correa Duarte, Miguel Ángel			
Pérez Juste, Jorge			
Silva López, Carlos			
macorrea@uvigo.es			
The course aims to familiarize students with the us	se of chemical inform	nation sources (	scientifical and technical
in general) with emphasis on its use through the Ir	nternet, as well as wi	ith the use of al	l types of software tools
communication skills (writing scientific and technic	cal documents, acad	emic, web desig	gn, etc).
	chemistry V11G200V01401 (*)Grao en Química  ECTS Credits 6 English  Correa Duarte, Miguel Ángel Correa Duarte, Miguel Ángel Pérez Juste, Jorge Silva López, Carlos macorrea@uvigo.es  The course aims to familiarize students with the usin general) with emphasis on its use through the Infor statistical calculations and chemical modeling	I communication in chemistry IT tools and communication in chemistry V11G200V01401 (*)Grao en Química  ECTS Credits Choose 6 Mandatory English  Correa Duarte, Miguel Ángel Correa Duarte, Miguel Ángel Pérez Juste, Jorge Silva López, Carlos macorrea@uvigo.es  The course aims to familiarize students with the use of chemical informin general) with emphasis on its use through the Internet, as well as wifor statistical calculations and chemical modeling . Attention is also pa	I communication in chemistry  IT tools and communication in chemistry  V11G200V01401  (*)Grao en Química  ECTS Credits Choose Year 6 Mandatory 2nd  English  Correa Duarte, Miguel Ángel Correa Duarte, Miguel Ángel Pérez Juste, Jorge Silva López, Carlos

Comp	Competencies		
Code			
C20	Evaluate, interpret and synthesize data and chemical information		
C22	Process and perform computational calculations with chemical information and chemical data		
C23	Present oral and written scientific material and scientific arguments to a specialized audience		
D1	Communicate orally and in writing in at least one of the official languages of the University		
D2	Communicate at a basic level in English in the field of chemistry		
D3	Learn independently		
D4	Search and manage information from different sources		
D5	Use information and communication technologies and manage basic computer tools		
D8	Teamwork		
D9	Work independently		
D10	Work at a national and international context		
D14	Analyze and synthesize information and draw conclusions		
D15	Evaluate critically and constructively the environment and oneself		
D16	Develop an ethical commitment		
D18	Generate new ideas and show initiative		

Learning outcomes		
Expected results from this subject	Training and Learning	
		Results
(*)Distinguish and handle the distinct sources of scientific and technical information (books,	C23	D1
magazines, summaries, databases, pages web, patents, etc.).		D2
		D4
		D5
		D9
		D14
		D16

(*) Differentiate and classify the scientific magazines and the contributions to the same, respect to	)	D2
their thematic, aim and scope.		D4
		D5
		D8
		D9 D14
(*) Find and absorb information in a fast and effective way.	C23	D14
( ) This and absolution matter and effective way.	023	D2
		D3
		D5
		D8
		D9
		D10
		D15
(*) Resume and classifiy the information for its effective broadcasting.	C23	D18 D1
(*) Resume and classify the information for its effective broadcasting.	C23	D1 D2
		D5
		D8
		D10
		D16
(*) Argue the own opinions showing critical sense.	C23	D1
		D2
		D5
		D8
		D10
(*) Performd simple written documents for the diffusion of knowledges and the scientific and	C23	D16 D1
technical results (p.ej. Articles, reports, works).	C23	D1 D2
technical results (p.ej. Articles, reports, works).		D5
		D8
		D10
		D16
(*) Handle with critical spirit the network (""""internet"""") as an information source.	C22	D3
		D5
		D9
		D14
(*) Defense and only and a second of the control of	622	D16
(*) Perform academic oral presentations on subjects related with the Chemistry, using audiovisual media.	C23	D1 D2
media.		D14
		D18
(*) Organise the bibliography, with or without help of bibliographic tools.	C20	D3
		D4
		D5
		D9
		D14
(*) Use computer programs for the preparation of figures and charts.	622	D15
(*) Use computer programs for the preparation of figures and charts.	C22	D4 D5
		D9
(*) Comprehend the basic principles and utility of simulation programs of chemical processes.	C22	D5
( )		D9
	_	D14
(*) Comprehend and explain texts in English related with Chemistry.	C23	D1
		D2
		D3
(4) Dorft stands de consente and conference de characteristics to Football and the conference of the c	622	D8
(*) Draft simple documents and perform short oral presentations in English, on subjects related with Chemistry.	C23	D1
with Chemistry.		D2 D3
		D8
		D14
(*) Identify the most important programs of molecular modelling and understand the usefulnes of	C20	D3
(*) Identify the most important programs of molecular modelling and understand the usefulnes of the results obtained.	C20	D3 D4

Contents

The scietific literature: general aspects.	Structure and classification of the literature.
	General rules of a literature search.
	Function, organization and use of a scientific library.
nformation Sources	Books.
miorination sources	Journals.
	Technical reports.
	Conference Proceedings.
	Patents.
	Thesis.
	Government Publications.
	Standards.
	Videos.
	Dictionaries.
	Directories
	Encyclopedias Databases
Using Internet	Basic Internet services.
	Remote connection and file transfer utilities.
	Remote Connection and the transfer difficies.
	Search engines.
	Electronic lists and subscription services.
	Other services.
	Structure, function and design of web pages.
Indexing and abstracting services	Identification of a scientific paper.
	The ISI Web of Knowledge (WOK).
	The Chemical Abstract Service (CAS) and the Scifinder.
	Other abstracting services.
	Handbooks.
Bibliographic Managers	Classification of bibliographic references: general principles.
	Use of popular software packages:
	Defends and Endade as assessed as
Droporation of a scientific tackging or accelerate	Refworks and Endnote as examples.
Preparation of a scientific, technical or academic document	raits of a scientific document.
uocument	References, tables and figures : general principles.
	Use of computer templates.
	Conoral acposts of the scientific style and the use of Facility
	General aspects of the scientific style and the use of English.

Planning			
	Class hours	Hours outside the classroom	Total hours
Master Session	14	28	42
Practice in computer rooms	26	52	78
Troubleshooting and / or exercises	2	22	24
Long answer tests and development	1.5	4.5	6

documents.

How to write: CVs, progress reports, grant requests and other academic

\*The information in the planning table is for guidance only and does not take into account the heterogeneity of the students.

Methodologies	
	Description
Master Session	The theoretical aspects of the subject are presented

Practice in computer Computer lab exercises: literature searches, use of bibliographic managers, use of statistical packages, report writing.

Troubleshooting and / or Report or article writing in English language.
exercises Simple exercises with modelling software

Description
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Assessment				
	Description	Qualification		nd Learning sults
Practice in computer rooms	Typically, literature searches	20	C22 C23	D1 D2 D3 D4 D5 D9 D15 D16
Troubleshooting and / or exercises	Tipically, database searches and use of utilities of modelling software.	40	C22 C23	D1 D2 D3 D4 D5 D8 D10 D14 D15 D18
Long answer tests and developmen	tWritten exam consisting of short questions.	40	•	D1 D2 D14 D15

## Other comments on the Evaluation

Attendance at practical lectures (seminars) is compulsory. The student will be given a rating (0-10) as long as he/she has attended 3 or more seminar sessions, has delivered at least two reports on the exercises or practices proposed by the teacher or has done a written exam.

If the student fails in the first call he/she will be asked to improve some of the exercises or perform new ones provided by the teacher. In addition he/she will have to undergo a more thorough exam, which will weight 50% of the final grade.

Sources of information
Basic Bibliography
Complementary Bibliography
Douville, J.A., The literature of chemistry, 1st,
Kaplan, S.M., The English-Spanish Spanish-English dictionary of chemistry, 2a,
Day, R.A.; Gastel, B., How to write and publish a scientific paper, 7 <sup>a</sup> ,

### Recommendations

#### Subjects that are recommended to be taken simultaneously

Numerical methods in chemistry/V11G200V01402 Physical chemistry II/V11G200V01403 Inorganic chemistry I/V11G200V01404

### Subjects that it is recommended to have taken before

Physics: Physics I/V11G200V01102
Physics: Physics II/V11G200V01201
Chemistry: Chemistry I/V11G200V01105
Chemistry: Chemistry 2/V11G200V01204

